

## Novel Stereospecific Synthesis of ( $\pm$ )-Dethiobiotin

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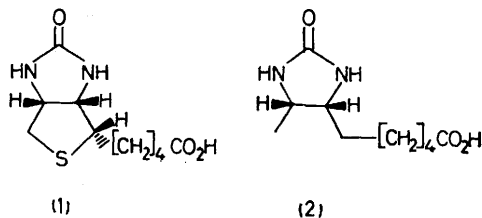
**Summary** A new stereospecific synthesis of dethiobiotin (**2**), the biological precursor of the vitamin biotin (**1**), is described.

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THE immediate precursor of the vitamin (+)-biotin (**1**) in a number of microbial systems has been found<sup>1</sup> to be (+)-dethiobiotin (**2**). As part of an investigation of biotin formation and metabolism, we attempted to prepare several

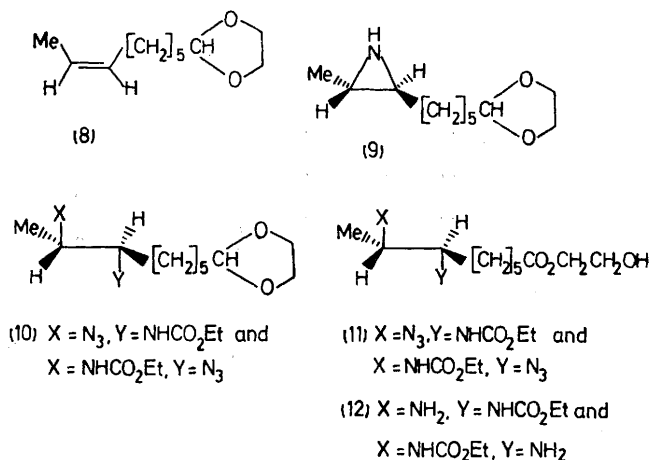
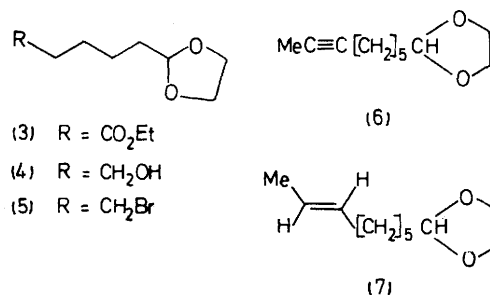
specifically tritiated forms of dethiobiotin by using published syntheses<sup>2</sup> of this compound. However, these attempts led to scrambling of the tritium labels. We have therefore developed a new synthesis of ( $\pm$ )-dethiobiotin which will be useful for labelling purposes. This approach differs from prior syntheses of (**2**) in that the *cis* relationship between the substituents on the imidazolone ring is achieved by means other than the catalytic reduction of an unsaturated imidazolone.

Rosemund reduction of 5-ethoxycarbonyl-n-pentanoyl chloride<sup>3</sup> followed by immediate acetalization of the ester aldehyde yielded the acetal ester (3) (62%),  $\nu_{\max}$  1740  $\text{cm}^{-1}$ ,  $\delta$  1.22 (3H, t,  $J$  7 Hz,  $\text{OCH}_2\text{Me}$ ), 4.12 (2H, q,  $J$  7 Hz,



$\text{OCH}_2\text{Me}$ ), and 4.85 (1H, t,  $J$  4.3 Hz, acetal-H). Treatment of (3) with  $\text{LiAlH}_4$  afforded the acetal alcohol (4) (90%) which was converted by  $\text{PPh}_3$  and  $\text{CBr}_4$  into the bromoacetal (5) (75%),  $\delta$  3.41 (2H, t,  $J$  6.5 Hz,  $\text{CH}_2\text{Br}$ ) and 4.82 (1H, t, acetal-H). Alkylation of propyne with (5) produced the acetylenic acetal (6) (60%),  $\delta$  1.73 (3H, t,  $J$  2.5 Hz,  $\text{MeC}\equiv\text{CCH}_2$ ) and 4.80 (1H, t, acetal H).  $\text{Na-NH}_3$  reduction of (6) led to the *trans*-acetal olefin (7) (93%) with an i.r. absorption (965  $\text{cm}^{-1}$ ) characteristic of a *trans*-substituted double-bond.<sup>4</sup> In order to provide additional evidence for the *trans* geometry of (7), the *cis*-acetal olefin (8) was synthesized by reduction (Lindlar) of the acetylene (6). The *cis*-olefin (8) exhibited i.r. absorption of 700  $\text{cm}^{-1}$  as expected for a *cis*-substituted double-bond,<sup>4</sup> and g.l.c. analysis of the *trans*-olefin (7) shows that none of the *cis*-olefin (8) was present. The *trans*-acetal olefin was subsequently used as the olefinic component of a stereospecific aziridine synthesis devised by Hassner.<sup>5</sup> Sequential treatment of (7) with  $\text{IN}_3$ ,  $\text{P}(\text{OEt})_3$ , and  $\text{LiAlH}_4$  afforded the *trans*-substituted aziridine (9) (70% from 7),  $\delta$  1.17 (3H, d,  $J$  5 Hz, aziridinyl-Me) and 4.80 (1H, t, acetal-H). Ring-opening of the aziridine with  $\text{NaN}_3$  in acidic, aqueous alcohol<sup>6</sup> followed by acylation with  $\text{ClCO}_2\text{Et}$  yielded a mixture of two isomeric azido-urethanes (10) (85%),  $\nu_{\max}$  2110 and 1725  $\text{cm}^{-1}$ ,  $\delta$  1.25 (3H, t,  $J$  7 Hz,  $\text{OCH}_2\text{Me}$ ) and 4.12 (2H, q,  $J$  7 Hz,  $\text{OCH}_2\text{Me}$ ). Ozonolysis<sup>7</sup> of the mixture (10) at  $-78^\circ\text{C}$  in MeOH cleanly produced a mixture of two isomeric ethylene glycol esters (11),  $\nu_{\max}$  2110, 1725, and 1715  $\text{cm}^{-1}$ ,  $\delta$  2.31 (2H, t,  $J$  7 Hz,  $\text{CH}_2\text{CO}_2$ ), and 3.80 (2H, t,  $\text{OCH}_2\text{CH}_2\text{OH}$ ). Catalytic reduction of

the ester mixture (11) gave a mixture of two isomeric amino esters (12) which was treated with  $\text{NaOEt}$  in EtOH to afford



the ethyl ester of ( $\pm$ )-dethiobiotin (60% from 10). Alkaline hydrolysis of this ester yielded crystalline ( $\pm$ )-dethiobiotin whose identity was established by comparison with a sample of ( $\pm$ )-dethiobiotin synthesized *via* previously published methods. The two samples of ( $\pm$ )-(2) were identical in all respects, including mixed m.p.

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<sup>1</sup> See H.-C. Li, D. B. McCormick, and L. D. Wright, *J. Biol. Chem.*, 1968, **243**, 6442, and the references cited therein.

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<sup>7</sup> P. Deslongchamps and C. Moreau, *Canad. J. Chem.*, 1971, **49**, 2465.